

Aluminum Plasma Etch Guide in the Trion Metal Etcher

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Disclaimer: The information below is to act as a starting point for etching your aluminum. Expect variation in etch rates dependent on area etched, feature sizes and the precise details of your sample.

1. Contact clean room staff before performing the etch to make sure BCl_3 and Cl_2 lines are charged. Have staff set the chiller to 35°C to prevent BCl_3 condensation and all BCl_3 lines are heated.
2. General notes concerning chlorine based etching of aluminum¹⁻⁵
 - Aluminum forms a native oxide that is not etched by pure chlorine chemistry thus BCl_3 is used to break through the oxide.
 - The general etch byproduct is AlCl_3
 - AlCl_3 is hygroscopic and the material is not very volatile
 - AlCl_3 will react with water vapor, resist and pump oil
 - BCl_3 is added to the plasma to remove the aluminum oxide and scavenge oxides formed from residual water or oxygen
 - Resist exposed to BCl_3 or Cl_2 will become resistant to oxygen plasma etching or solvation after they've been exposed to air/water vapor. This may be due to the conversion of AlCl_3 to Al_2O_3 .
 - Aluminum alloys such as Al-Si-Cu or Al-Si have significantly different etch properties than pure aluminum.
 - Addition of He to the plasma may help to improve resist selectivity
 - If the percentage of Cl_2 is lowered to 5% in a mixture with BCl_3 the lateral etch rate will be reduced. ¹
 - Addition of NH_3 may also decrease undercut⁶

3. Aluminum silicon etching

The recipe for plasma etching aluminum is dependent on the type and method of aluminum deposition. For example, Al-Si 1% etches much slower than pure aluminum and electron beam evaporated Al probably etches at a different rate than sputtered metal.

The recipe for etching Al-Si 1% occurs in 2 steps. The first etches the aluminum and the second step passivates the photo resist (which is loaded with chlorine) from ambient water vapor. The passivation step helps to ensure that the photo resist can be removed in

common organic solvents such as acetone or NMP. Note that other recipes indicate that He can be used to improve selectivity to photoresist⁴. Unfortunately, the CF₄ will also etch silicon oxide, thus we minimize the plasma time and in future recipes it may be helpful to minimize DC bias for this step. The high bias voltage in the first step helps to remove redeposition of the silicon in the metal layer. Further, it is necessary to run the chuck holding the sample at 35°C to prevent condensation of BCl₃ and etch byproducts.

Recipe: Al_Best_Etch_ALSi

Etch Rate: 3.89 nm/sec, 350 nm in 90 sec

Step 1 Etch Step

Chiller Temperature set to 35°C

BCl₃: 15 sccm Cl₂: 35 sccm, Pressure: 20 mTorr

ICP: 350W RFL: <25, RIE 250 RFL: <25 DC BIAS: -400V to -500 V

He Cooling: 5 Torr

Step 2 Passivation Step

CF₄: 40 sccm, Pressure: 25 mTorr

ICP RF:400 W, RIE RF:100 W, DC Bias: -125 V

He Cooling: 0 Torr

Time: 20 to 40 seconds

Or

O₂: 50 sccm, Press: 50 mT

ICP: 300W+-25 RFL: <10 RIE: 75W+-25 RFL: <15 DC Bias: ~- 100V

He Cooling: 0 Torr

Time 120 sec

It may also be possible to remove the resist with an oxygen plasma etch as long as the sample has not been exposed to air.

4. Pure aluminum etching

The recipe for etching pure aluminum is less aggressive than the aluminum-silicon etch as a high bias voltage is used to prevent silicon redeposition on the etched area.

Recipe: Al_Best_Etch

Etch Rate: 12.08 nm/sec

Step 1

BCl₃: 15 sccm Cl₂: 35 sccm, Pressure: 20 mTorr

ICP: 350W RFL: <25, RIE 150 RFL: <25 DC BIAS: -400V to -500 V

He Cooling: 5 Torr

Step 2

CF₄: 40 sccm, Pressure: 25 mTorr

ICP RF:400 W, RIE RF:100 W, DC Bias: -125 V

He Cooling: 0 Torr

Time: 20 to 40 seconds

Or

O₂: 50 sccm, Press: 50 mT
ICP: 300W+-25 RFL: <10 RIE: 75W+-25 RFL: <15 DC Bias: ~- 100V
He Cooling: 0 Torr
Time 120 sec

5. Etching Al-Si-Cu

Etching of this common alloy may require a second wet etch to remove traces of copper as compounds of copper with fluorine and chlorine are nonvolatile.

References

1. Hess, D. W., Plasma Etch Chemistry of Al and Al Alloy Films. *Plasma Chemistry and Plasma Processing* **1982**, 2 (2), 141-155.
2. Fuller, L., Plasma Etching 2013.
3. Donnelly, V. M.; Kornblit, A., Plasma etching: Yesterday, today, and tomorrow. *Journal of Vacuum Science & Technology A: Vacuum, Surfaces, and Films* **2013**, 31 (5).
4. Bruce, R. H.; Malafsky, G. P., High Rate Anisotropic Aluminum Etching *Journal of Vacuum Science & Technology* **1983**, 130 (6), 1369-1373.
5. Frank, W. E., Approaches for patterning of aluminum. *Microelectronic Engineering* **1997**, 33, 85-100.
6. Yamanaka, M. Dry Etching Method for for Aluminum Alloy Etching Gas Therefor. 5,837,616, 1998.