GSTROM SCIEN

40 S. Linden Street, Duquesne, PA 15110 USA

T: +1-412-469-8466

F; +1-412-469-8511

angstromsciences.com

MATERIAL SAFETY DATA SHEET COMPLIES WITH 29 CFR 1910.1200. **OSHA HAZARD COMMUNICATION RULE**

DATE OF LAST REVISION: 12-06-04

CHEMICAL IDENTITY

PRODUCT NAME SYNONYMS

FORMULA CHEMICAL FAMILY CAS REGISTRY NUMBER HAZARDOUS INGREDIENTS **IRON (III) OXIDE** FERRIC OXIDE, RED IRON OXIDE, IRON RED, **IRON SESQUIOXIDE** Fe₂O₃ **METAL OXIDE** 1309-37-1 PEL - 10mg/m^3 TLV - 5mg/m³ % - 100%

PHYSICAL AND CHEMICAL PROPERTIES

COLOR, FORM AND ODOR **BOILING POINT** VAPOR DENSITY (air=1) VAPOR PRESSURE % VOLATILE BY VOLUME (%) **REACTION WITH WATER** EVAPORATION RATE (butyl acetate=1) SOLUBILITY IN WATER FREEZING/MELTING POINT

Reddish powder/target; odorless NA NA NA NA NA NA Insoluble NA

FIRE AND EXPLOSION HAZARD DATA

FLASH POINT AUTOIGNITION TEMPERATURE (°C) FLAMMABILITY **EXTINGUISHING MEDIA**

NA NA NA Use fire fighting measures that suit the surrounding fire. Do not use water or carbon dioxide as extinguishing agent.



40 S. Linden Street, Duquesne, PA 15110 USA

T: +1-412-469-8466

F; +1-412-469-8511 angstromsciences.com

IRON OXIDE MATERIAL SAFETY DATA SHEET

FIRE AND EXPLOSION HAZARD DATA CONTINUED

UNUSUAL FIRE & EXPLOSION HAZARDS May have a violent reaction when heated with powdered aluminum, calcium disilicide, magnesium, and metal acetylides. Reacts explosively when heated with guanidinium perchlorate. Iron (III) Oxide has the potential to react violently with hydrogen peroxide and reaction with carbon monoxide may form an explosive product. The wet oxide may react explosively with molten aluminum-magnesium alloy.

HEALTH HAZARD INFORMATION

MEDICAL CONDITIONS AGGRAVATED BY OVEREXPOSURE: ND

EFFECTS OF OVEREXPOSURE (acute):

Ingestion: None known

Skin contact: None known

Eye contact: May cause irritation.

Inhalation: May irritate upper respiratory tract. May cause acute iron poisoning. Acute iron poisoning may cause biphasic shock, a rapid increase in respiration and pulse rate, congestion of blood vessels which may lead to hypotension, pallor and drowsiness. Other health hazards: None known

EFFECTS OF OVEREXPOSURE (chronic):

Ingestion: None known

Skin contact: Repeated or prolonged contact may cause irritation.

Eye contact: None Known

Inhalation: Prolonged exposure to dust may cause benign pneumoconiosis. May cause chronic iron poisoning. Effects include hemorragic necrosis of the gastrointestinal tract, hepatotoxicity, metabolic acidosis, prolonged blood clotting time and elevation of plasma levels or serotonin and histamine.



ANGSTROM SCIENCES

40 S. Linden Street, Duquesne, PA 15110 USA

T: +1-412-469-8466 F; +1-412-469-8511

angstromsciences.com

IRON OXIDE

MATERIAL SAFETY DATA SHEET

CARCINOGENICITY: No NTP: No IARC MONOGRAPHS: No OSHA REGULATE: No

EMERGENCY FIRST AID PROCEDURES:

INGESTION: Give at last 2 glasses of milk or water. Induce vomiting unless patient is unconscious. Call a physician.

INHALATION: Remove to fresh air and give artificial respiration if not breathing. SKIN CONTACT: Wash thoroughly with soap and water.

EYE CONTACT: Immediately flush eyes, including under eyelids, with large amounts of water for at least 15 minutes. Call a physician.

REACTIVITY DATA

STABILITY: Stable

INCOMPATIBILITY (MATERIALS TO AVOID): May have a violent reaction when heated with powdered aluminum, calcium disilicide, magnesium, and metal acetylides. Reacts explosively when heated with guanidinium perchlorate. Iron (III) Oxide has the potential to react violently with

hydrogen peroxide and reaction with carbon monoxide may forma an explosive product. The wet oxide may react explosively with molten aluminum-magnesium alloys.

HAZARDOUS DECOMPOSITION PRODUCTS: ND HAZARDOUS POLYMERIZATION: Will Not Occur CONDITIONS TO AVOID: Incompatibles OTHER: ND

SAFE HANDLING AND USE

RCRA Code: D010 TSCA Registered: Yes

SPILL AND LEAK PROCEDURES: Wear protective clothing to protect against hazards such as overexposure to skin, eyes, and respiratory tract. Collect all material which may be rleased, into containers, label, and hold for proper disposal per local, state, and federal regulations.

WASTE DISPOSAL METHOD: Consult state, local or federal EPA regulations

SPECIAL PROTECTIVE INFORMATION

VENTILATION REQUIREMENTS: Laboratory fume hood or sufficient ventilation to ensure exposure remains below TLV/PEL.

RESPIRATORY PROTECTION: If there is dustiness, wear a respirator selected per OSHA standards.

PROTECTIVE GLOVES: Rubber



40 S. Linden Street, Duquesne, PA 15110 USA

T: +1-412-469-8466 F; +1-412-469-8511

angstromsciences.com

IRON OXIDE MATERIAL SAFETY DATA SHEET

EYE PROTECTION: Splash goggles, safety glasses, or face shields recommended. OTHER PROTECTIVE EQUIPMENT: An apron, or other impermeable body protection suggested. Wash any contaminated clothing before reuse.

SPECIAL PRECAUTIONS

HANDLING/STORAGE: Keep container tightly closed. Store in a cool, dry, well-ventilated area. Wash thoroughly after use.

OTHER PRECAUTIONS: Lab coat and apron, flame and chemical resistant coveralls, eyewash capable of sustained flushing, safety drench shower and hygenic facilities for washing.

WARNING: Dust may irritate eyes and respiratory tract.

TRANSPORTATION REQUIREMENTS

DOT CLASS: NA UN NUMBER: NA IMCO CLASS: NA OTHER: NA

PRECAUTIONARY LABELING: This product contains Selenium which is subject to the reporting requirements of section 313 of the Emergency Planning and Community Right-to-know Act of 1986 and 40CFR 372.

THE ABOVE INFORMATION IS ACCURATE TO THE BEST OF OUR KNOWLEDGE. HOWEVER, SINCE DATA, SAFETY STANDARDS AND GOVERNMENT REGULATIONS ARE SUBJECT TO CHANGE THE CONDITIONS OF HANDLING AND USE, OR MISUSE ARE BEYOND OUR CONTROL, ANGSTROM SCIENCES MAKE NO WARRANTY, EITHER EXPRESSED OR IMPLIED, WITH RESPECT TO THE COMPLETENESS OR CONTINUING ACCURACY OF THE INFORMATION CONTAINED HEREIN AND DISCLAIMS ALL LIABILITY FOR THE RELIANCE THEREON. USER SHOULD SATISFY HIMSELF THAT HE HAS ALL CURRENT DATA RELEVANT TO HIS PARTICULAR USE. NA= NOT APPLICABLE

ND= NO DATA FOUND