

University of Louisville  
Department of Chemistry  
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**Literature Seminar**

When: April 18, 2024

Time: 1:30 p.m.

Location: CBL-16

## **Recent strategies for electrocatalytic C-N bond formation for urea synthesis**

**Abstract:** Urea ( $\text{CO}(\text{NH}_2)_2$ ) is utilized in nitrogen-based fertilizers for agriculture. Electrocatalytic urea synthesis has gained traction as an alternative technology to the conventional energy-demanding industrial urea synthesis method.<sup>1</sup> However, innovative approaches are required to boost the electrocatalytic C-N coupling process and impede the competing reactions. In this presentation, I will discuss recent electrocatalyst design strategies and the current mechanistic understanding of electrocatalytic C-N coupling. Huang et al. reacted  $\text{CO}_2$  and NO in water at monometallic zinc nanobelts for electrocatalytic urea synthesis. Characterization and theoretical simulations demonstrated that the C-N bond formation originated from coupling  $^*\text{CO}$  and  $^*\text{NH}_2$  intermediates. However, the overall efficiency for urea synthesis was modest with a Faradaic efficiency (F.E) of 11.26% at -0.92 V vs RHE.<sup>2</sup> Zhang et al. proposed an approach to enhance the adsorption of  $\text{N}_2$  and  $\text{CO}_2$  molecules on the surface of ZnMn-N, Cl electrocatalysts. The authors used Mn to help elongate the  $\text{N}\equiv\text{N}$  bond to couple with a  $^*\text{CO}$  intermediate to realize C-N coupling in a single step in the electrosynthesis of urea. Limited by the bond activation of  $\text{N}_2$ , a moderate urea production rate and a F.E of 63.5% at -0.3 V vs RHE was achieved.<sup>3</sup> Zhao et al. utilized a  $\text{CuWO}_4$  electrocatalyst and  $\text{NO}_3^-$  as the nitrogen source to synthesize urea. To minimize C-C coupling, the catalyst was designed with alternating bimetallic sites, producing a high F.E of 70% at -0.2 V vs RHE. Mechanistic studies revealed  $^*\text{NO}_2$  and  $^*\text{CO}$  as the primary intermediates which likely amplified C-N coupling at a low overpotential.<sup>1</sup> Based on the knowledge of this literature review, a perspective on the future of C-N coupling in urea electrosynthesis will be given.

### **References:**

- (1) Zhao, Y.; Ding, Y.; Li, W.; Liu, C.; Li, Y.; Zhao, Z.; Shan, Y.; Li, F.; Sun, L.; Li, F. Efficient Urea Electrosynthesis from Carbon Dioxide and Nitrate via Alternating Cu–W Bimetallic C–N Coupling Sites. *Nat. Commun.* **2023**, *14* (1), 4491.
- (2) Huang, Y.; Yang, R.; Wang, C.; Meng, N.; Shi, Y.; Yu, Y.; Zhang, B. Direct Electrosynthesis of Urea from Carbon Dioxide and Nitric Oxide. *ACS Energy Lett.* **2022**, *7* (1), 284–291.
- (3) Zhang, X.; Zhu, X.; Bo, S.; Chen, C.; Cheng, K.; Zheng, J.; Li, S.; Tu, X.; Chen, W.; Xie, C.; Wei, X.; Wang, D.; Liu, Y.; Chen, P.; Jiang, S. P.; Li, Y.; Liu, Q.; Li, C.; Wang, S. Electrocatalytic Urea Synthesis with 63.5 % Faradaic Efficiency and 100 % N-Selectivity via One-step C–N Coupling. *Angew. Chem. Int. Ed.* **2023**, *62* (33), e202305447.