MATERIAL SAFETY DATA SHEET
OSHA HAZARD COMMUNICATION RULE

DATE OF LAST REVISION: 12-06-04

CHEMICAL IDENTITY

<table>
<thead>
<tr>
<th>PRODUCT NAME</th>
<th>IRON (III) OXIDE</th>
</tr>
</thead>
<tbody>
<tr>
<td>SYNONYMS</td>
<td>FERRIC OXIDE, RED IRON OXIDE, IRON RED, IRON SESQUIOXIDE</td>
</tr>
<tr>
<td>FORMULA</td>
<td>Fe$_2$O$_3$</td>
</tr>
<tr>
<td>CHEMICAL FAMILY</td>
<td>METAL OXIDE</td>
</tr>
<tr>
<td>CAS REGISTRY NUMBER</td>
<td>1309-37-1</td>
</tr>
<tr>
<td>HAZARDOUS INGREDIENTS</td>
<td>PEL - 10mg/m$^3$</td>
</tr>
<tr>
<td></td>
<td>TLV - 5mg/m$^3$</td>
</tr>
<tr>
<td></td>
<td>% - 100%</td>
</tr>
</tbody>
</table>

PHYSICAL AND CHEMICAL PROPERTIES

<table>
<thead>
<tr>
<th>COLOR, FORM AND ODOR</th>
<th>Reddish powder/target; odorless</th>
</tr>
</thead>
<tbody>
<tr>
<td>BOILING POINT</td>
<td>NA</td>
</tr>
<tr>
<td>VAPOR DENSITY (air=1)</td>
<td>NA</td>
</tr>
<tr>
<td>VAPOR PRESSURE</td>
<td>NA</td>
</tr>
<tr>
<td>% VOLATILE BY VOLUME</td>
<td>NA</td>
</tr>
<tr>
<td>REACTION WITH WATER</td>
<td>NA</td>
</tr>
<tr>
<td>EVAPORATION RATE</td>
<td>NA</td>
</tr>
<tr>
<td>SOLUBILITY IN WATER</td>
<td>Insoluble</td>
</tr>
<tr>
<td>FREEZING/MELTING POINT</td>
<td>NA</td>
</tr>
</tbody>
</table>

FIRE AND EXPLOSION HAZARD DATA

| FLASH POINT           | NA                              |
| AUTOIGNITION TEMPERATURE (°C) | NA                              |
| FLAMMABILITY         | NA                              |
| EXTINGUISHING MEDIA  | Use fire fighting measures that suit the surrounding fire. Do not use water or carbon dioxide as extinguishing agent. |
IRON OXIDE
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FIRE AND EXPLOSION HAZARD DATA CONTINUED

UNUSUAL FIRE & EXPLOSION HAZARDS
May have a violent reaction when heated with powdered aluminum, calcium disilicide, magnesium, and metal acetylides. Reacts explosively when heated with guanidinium perchlorate. Iron (III) Oxide has the potential to react violently with hydrogen peroxide and reaction with carbon monoxide may form an explosive product. The wet oxide may react explosively with molten aluminum-magnesium alloy.

HEALTH HAZARD INFORMATION

MEDICAL CONDITIONS AGGRAVATED BY OVEREXPOSURE: ND

EFFECTS OF OVEREXPOSURE (acute):
- Ingestion: None known
- Skin contact: None known
- Eye contact: May cause irritation.
- Inhalation: May irritate upper respiratory tract. May cause acute iron poisoning. Acute iron poisoning may cause biphasic shock, a rapid increase in respiration and pulse rate, congestion of blood vessels which may lead to hypotension, pallor and drowsiness.
- Other health hazards: None known

EFFECTS OF OVEREXPOSURE (chronic):
- Ingestion: None known
- Skin contact: Repeated or prolonged contact may cause irritation.
- Eye contact: None known
- Inhalation: Prolonged exposure to dust may cause benign pneumoconiosis. May cause chronic iron poisoning. Effects include hemorrhagic necrosis of the gastrointestinal tract, hepatotoxicity, metabolic acidosis, prolonged blood clotting time and elevation of plasma levels or serotonin and histamine.
EMERGENCY FIRST AID PROCEDURES:

INGESTION: Give at least 2 glasses of milk or water. Induce vomiting unless patient is unconscious. Call a physician.

INHALATION: Remove to fresh air and give artificial respiration if not breathing.

SKIN CONTACT: Wash thoroughly with soap and water.

EYE CONTACT: Immediately flush eyes, including under eyelids, with large amounts of water for at least 15 minutes. Call a physician.

REACTIVITY DATA

STABILITY: Stable

INCOMPATIBILITY (MATERIALS TO AVOID): May have a violent reaction when heated with powdered aluminum, calcium disilicide, magnesium, and metal acetylides. Reacts explosively when heated with guanidinium perchlorate. Iron (III) Oxide has the potential to react violently with hydrogen peroxide and reaction with carbon monoxide may form an explosive product. The wet oxide may react explosively with molten aluminum-magnesium alloys.

HAZARDOUS DECOMPOSITION PRODUCTS: ND
HAZARDOUS POLYMERIZATION: Will Not Occur
CONDITIONS TO AVOID: Incompatibles
OTHER: ND

SAFE HANDLING AND USE

RCRA Code: D010
TSCA Registered: Yes

SPILL AND LEAK PROCEDURES: Wear protective clothing to protect against hazards such as overexposure to skin, eyes, and respiratory tract. Collect all material which may be released, into containers, label, and hold for proper disposal per local, state, and federal regulations.

WASTE DISPOSAL METHOD: Consult state, local or federal EPA regulations

SPECIAL PROTECTIVE INFORMATION

VENTILATION REQUIREMENTS: Laboratory fume hood or sufficient ventilation to ensure exposure remains below TLV/PEL.
RESPIRATORY PROTECTION: If there is dustiness, wear a respirator selected per OSHA standards.
PROTECTIVE GLOVES: Rubber
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EYE PROTECTION: Splash goggles, safety glasses, or face shields recommended.
OTHER PROTECTIVE EQUIPMENT: An apron, or other impermeable body protection suggested.
Wash any contaminated clothing before reuse.

SPECIAL PRECAUTIONS

HANDLING/STORAGE: Keep container tightly closed. Store in a cool, dry, well-ventilated area. Wash thoroughly after use.

OTHER PRECAUTIONS: Lab coat and apron, flame and chemical resistant coveralls, eyewash capable of sustained flushing, safety drench shower and hygienic facilities for washing.

WARNING: Dust may irritate eyes and respiratory tract.

TRANSPORTATION REQUIREMENTS

DOT CLASS: NA
UN NUMBER: NA
IMCO CLASS: NA
OTHER: NA

PRECAUTIONARY LABELING: This product contains Selenium which is subject to the reporting requirements of section 313 of the Emergency Planning and Community Right-to-know Act of 1986 and 40CFR 372.

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NA= NOT APPLICABLE
ND= NO DATA FOUND